Genie in a Bottle: Decomposition of Hydrogen Peroxide

Instructions

Materials: 30% Hydrogen peroxide, H₂O₂ 2-liter plastic soft drink bottle Manganese dioxide, MnO₂ Ring stand with ring clamp

<u>Safety</u>: 30% hydrogen peroxide is corrosive. Avoid contact with skin and eyes. Use safety glasses.

Set Up:

- 1. Hold the 2-liter plastic soft drink bottle in place by slipping a ring over the bottle.
- 2. Take the bottle of 30% hydrogen peroxide out of refrigerator where it's normally stored.
- 3. Measure out 3-4 g of manganese dioxide. The fine powder solid works best.

Demo Procedure:

- 1. Carefully pour 50 mL of 30% hydrogen peroxide into the bottle.
- 2. Add manganese dioxide into the bottle and step away quickly.
- 3. The react is immediate and highly exothermic. A visible cloud of steam shoots out of the bottle.
- 4. The heat released can cause the bottle to shrink. The ring clamp helps keep the bottle in place.
- 5. Manganese dioxide is acting as a catalyst in this reaction, which means it can be used again. To demonstrate this point you can pour this content into a second bottle (with fresh hydrogen peroxide) and observe a similar reaction.

<u>Waste Disposal</u>: The solution, mostly water, can be pour down the drain. Pour the remaining wet MnO_2 solid into an evaporating dish or wide-mouth container and set it aside to dry. The dried MnO_2 can be reused multiple times.