**Rank** each set of acids in order of *increasing acidity*. The molecule numbered 1 is the least acidic and the molecule numbered 4 is most acidic. Then go back and **draw** the conjugate bases and **rank** those in order of *increasing basicity*.

$$N$$
 OH  $NH_4$ 

HO 
$$\stackrel{\circ}{\mathbb{S}}$$
  $-$  OH  $\stackrel{\circ}{\mathbb{S}}$   $-$  O