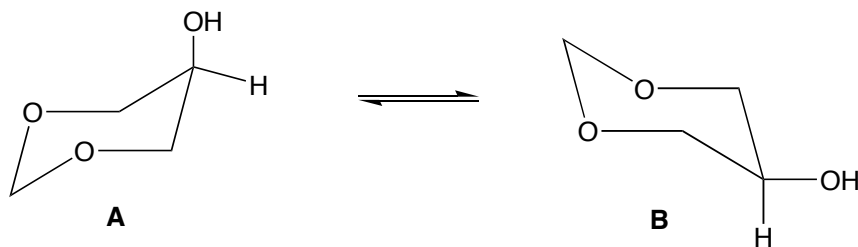


9. (10 pts) Draw structures of all the possible products of the monochlorination of 1-bromo-2-methylpropane. All products have the formula C_4H_8BrCl .

10. (5 pts) An intramolecular hydrogen bond plays a significant role in the conformational analysis of the compound below. **Circle the one** letter (a-e) which best describes the conformations of this molecule.



- a) Both **A** and **B** have the hydrogen bonding, and $K_{eq} > 1$.
- b) Only **B** has the hydrogen bonding, and $K_{eq} > 1$.
- c) Only **A** has the hydrogen bonding, and $K_{eq} < 1$.
- d) Only **B** has the hydrogen bonding, and $K_{eq} < 1$.
- e) Both **A** and **B** have the hydrogen bonding, and $K_{eq} = 1$.

$$K_{eq} = \frac{\left[\begin{array}{c} \text{B} \end{array} \right]}{\left[\begin{array}{c} \text{A} \end{array} \right]}$$